Q.P. Code: (19HS0801)



Reg. No:					

SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS)

B.Tech I Year II Semester Regular Examinations October-2020 APPLIED CHEMISTRY

(Common to CSE & CSIT)

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Time:	3 hours Max. Marks	s: 60							
	(Answer all Five Units $5 \times 12 = 60$ Marks)								
	UNIT-I								
1	a What is Electrochemical cell? Give an example.	6M							
	b Write a brief note on potentiometric sensor.	6M							
	OR								
2	a What is a Fuel cell? Describe the Construction and Working of Methanol–Oxygen Fuel cell.	8M							
	b Write a short note on Alkali metal sulphide batteries.	4M							
	UNIT-II								
3	a Explain the energy level diagrams of CO and NO molecule. Explain their magnetic	8M							
	nature and Bond order.								
	b Write the postulates of molecular orbital theory.	4M							
4	OR a Explain the application of Ψ and Ψ^2 to hydrogen atom.								
4	b Explain pi- molecular orbital's of Butadiene with a neat sketch.	6M							
	UNIT-III	6M							
5	a Distinguish between Thermoplastics and thermosetting plastics.	6M							
	b Explain the mechanism of cationic addition polymerization.	6M							
	OR								
6	a Describe the preparation, properties and uses of Nylon-6, 6.	5M							
	b Write the preparation, properties and uses of Phenol-Formaldehyde resin.	7M							
	UNIT-IV								
7	a Give an account on principle and instrumentation of IR spectroscopy.	7M							
	b Write a short note on Beer-Lambert's Law.	5M							
	OR								
8	a What is meant by Chromatography?	4M							
	b Explain the principle and instrumentation of Gas Chromatography.	8M							
	UNIT-V								
9	a Explain in detail about principle and application of semiconductors.	8M							
	b What is basic lock and key principle?	4M							
	OR								
10	a What is meant by Nanomaterials? How is Nanomaterials Classified?	5M							
	b Define Dielectrics. What are the characteristics of Electrical Insulators?	7M							

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